

266. Investigation on the Head-Space of Roasted Meat. III. Synthesis of 4,6-Dimethyl-2,3,5,7-tetrathiooctane¹⁾

by Paul Dubs²⁾ and Martin Joho

Givaudan Ltd, Research Company, CH-8600 Dübendorf-Zürich

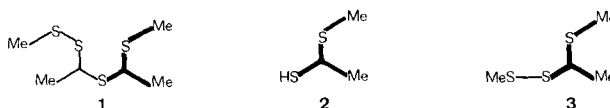
(10. VII. 78)

Summary

A synthesis of 4,6-dimethyl-2,3,5,7-tetrathiooctane (**1**), from 3,5-dimethyl-1,2,4-trithiolane (**6**), is described. Compound **1**, possessing a unique structure was recently found to be a constituent of roasted pork meat.

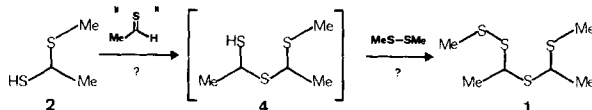
Introduction. - Analytical work performed in this laboratory by Hřivnáč *et al.* [1] has led to the assignment of structure **1** (NMR. and MS. data) to a compound occurring in the head-space of roasted pork meat. Compound **1** strongly resembles at least 2 structures **2** [2] and **3** [1] known from meat analyses (*Scheme 1*). The building-

Scheme 1



block **2** is incorporated in **3**, as well as in **1**. A reasonable pathway relating **2** and **3** was suggested by us [3]. Generically **1** could be the product of chain expansion of **2** with thioacetaldehyde or precursors thereof, leading to a hypothetical mercaptan intermediate **4** (*Scheme 2*), which could then be quenched in the same way as suggested for the formation of **3** from **2** [3].

Scheme 2

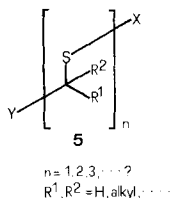


Structure **1** is the first isolated representative of the 2,3,5,7-tetrathiooctanes; however, it seems probable that less volatile homologues will be found in the future. This expectation is justified because various 3-substituted 2,4,5-trithiahexanes co-occur in roasted pork-meat [1], compound **3** being structurally the most simple example isolated.

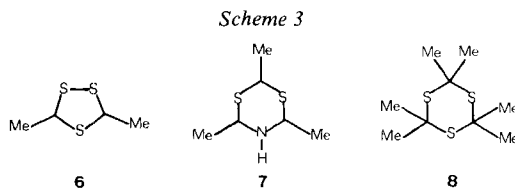
¹⁾ Part II: see [3].

²⁾ New address: *Uni-Chemie Ltd.*, Industriestr. 8, CH-8604 Volketswil-Zürich.

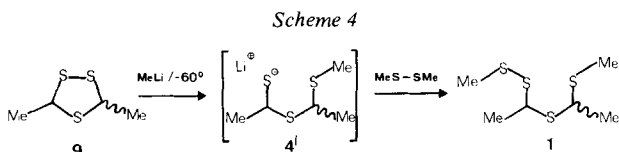
From a formal point of view thioaldehyde and thioketone derivatives of type **5** seem to play a dominant role in heat-treated, sulfur-containing food, such as roasted meat.



Nature seems to be mainly concerned with how to start (group Y) and how to terminate (group X) the n-fold sequence of thiocarbonyl-derived units as depicted in **5**. Besides compounds **1**, **2**, **3**, 3 further structures **6** [4], **7** [2] and **8** [5], known from meat-analyses, fitting very well into the framework **5** discussed, can be quoted; the latter examples being an illustration of the many possibilities realized for X and Y in nature.



Synthesis of 4,6-dimethyl-2,3,5,7-tetrathiaoctane (1). - Guided by 'biogenetic' reasoning, such as outlined in *Scheme 2*, we effected a short synthesis of **1** (*cf. Scheme 4*). Deprotonated 1-methylthiomercaptans are not too prone to degrade into their parts [3]. Therefore we focused our attention on the deprotonated form **4'** of the postulated intermediate **4**.



3,5-Dimethyl-1,2,4-trithiolane (**9**) [6] (*ca.* 60% *cis*- and 40% *trans*-isomers by NMR. assignments, *cf. Tjan et al.* [6]) reacted with methyllithium at -60° to give the anticipated intermediate **4'**, which was transformed to **1** in the presence of excess dimethyldisulfide. The product **1** is a mixture of diastereoisomers, readily separated by preparative GC. However, configurational assignment on the basis of their spectral data was not possible. The 3:2 ratio of the *cis* and *trans* starting material **9** seems to be reflected in a 3:2 ratio of isomeric pairs (\pm)-**1a** and (\pm)-**1b**. Based on the assumption that the chiral centers in intermediate **4'** remain configurationally stable under the reaction conditions, we tentatively assign the *erythro*-configuration to (\pm)-**1a** (originating from *cis*-**9**) and the *threo*-configuration to (\pm)-**1b** (originating from *trans*-**9**).

The NMR. spectra of the synthetic materials (\pm)-**1a** and (\pm)-**1b** and the isolated compound are illustrated. The isolate seems to consist of the *erythro*-form of **1** (the

small signal at $\delta = 2.2$ ppm is not conclusive evidence for the co-occurrence of the *threo*-isomer). However, in view of the delicate isolation technique [1] and since no argument is strongly in favour of only one isomer, both isomers (\pm)-**1a** and (\pm)-**1b** are probably present in roasted pork meat.

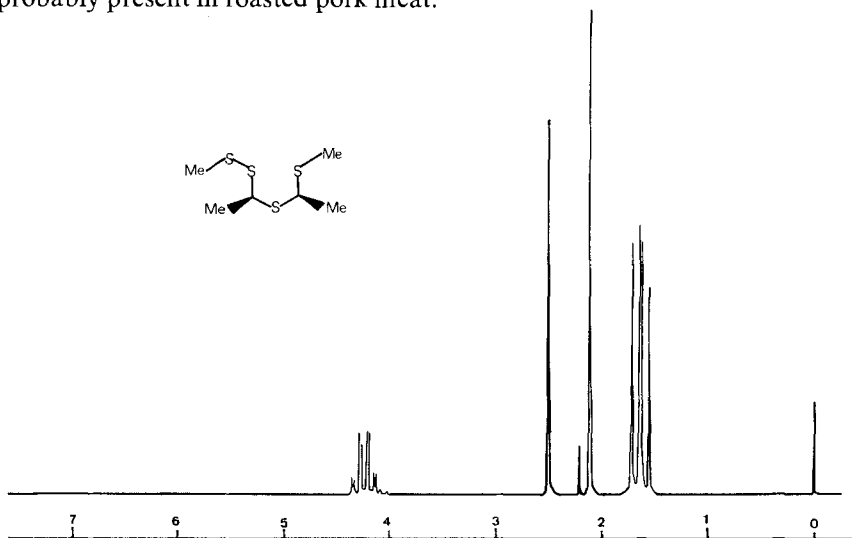


Fig. 1. $^1\text{H-NMR}$. of synthetic erythro (\pm)-**1a**

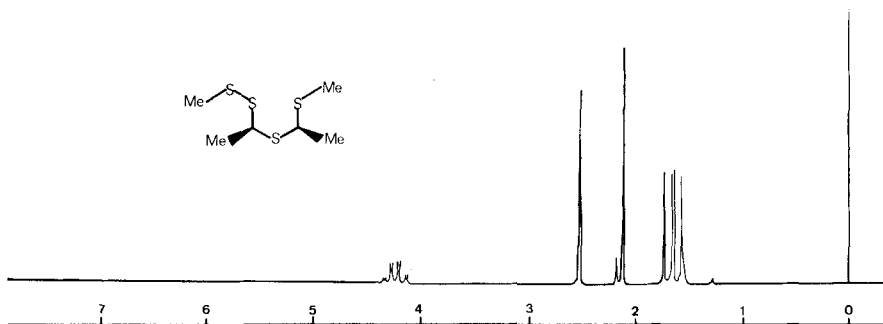
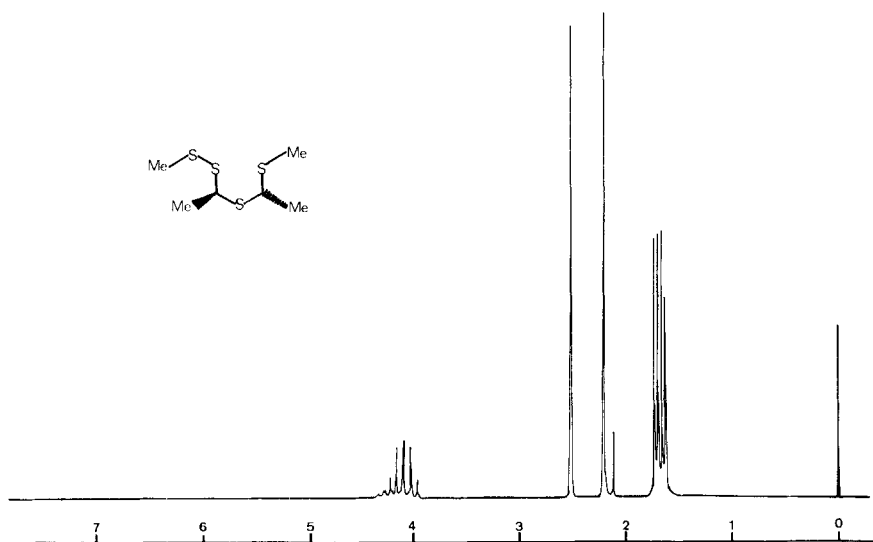


Fig. 2. $^1\text{H-NMR}$. of isolated, natural erythro (\pm)-**1a**

Experimental Part

General remarks. - $^1\text{H-NMR}$. spectra were recorded on a *Varian XL-100A* instrument (100 MHz), in CDCl_3 with TMS as internal standard (ppm values relative to TMS=0, J in Hz); abbreviations: s =singlet, d =doublet, qa =quartet. MS. were measured on a *Varian CH-5* spectrometer, using an inlet temperature of 150° and an ionisation energy of 70 eV; the intensity of the molecular ion and of the 8 most intense fragments' ions are given in % of the base peak. Gas liquid chromatography (GC.) was performed on a *Carlo Erba Fractovap GI* instrument, using OV-101, 2% on AW-DMCS, 80-100 mesh ($3\text{ mm} \times 3\text{ m}$).

3,5-Dimethyl-1,2,4-trithiolane (9) was prepared following [6] and was a mixture of *cis*- and *trans*-**9** 3:2.

Fig. 3. $^1\text{H-NMR}$. of synthetic threo (\pm)-**1b**

4,6-Dimethyl-2,3,5,7-tetrathiooctane (**1**). 3,5-Dimethyl-1,2,4-trithiolane (**9**) (3.04 g, 20 mmol) in 10 ml abs. ether was added dropwise within 10 min, under argon, to a well stirred 1.8M ethereal solution (11 ml) of freshly prepared CH_3Li at -60° . The solution was stirred at -60° for a further 90 min, then 50 ml of dimethyl disulfide were added with stirring at -50° and this temperature maintained for another 60 min. The cooling-bath was removed and stirring continued for 50 min to reach RT. The reaction mixture was washed to neutrality with saturated NaCl solution (6×25 ml). The organic layer was dried (Na_2SO_4) and evaporated at $60^\circ/11$ Torr. The crude product was subjected to a short-path distillation at 0.08 Torr/ 100° (oven temp.), giving a practically pure 3:2 mixture (3.6 g; 84%) of both possible diastereomers (\pm)-**1a** and (\pm)-**1b** (GC., NMR.), the more abundant isomer (\pm)-**1a** showing a distinctly shorter GC. retention on a packed OV-101 column. Pure (\pm)-**1a** and (\pm)-**1b** were obtained by preparative GC. on a OV-101 column.

Erythro-4,6-dimethyl-2,3,5,7-tetrathiooctane ((\pm))-**1a**. $^1\text{H-NMR}$.: 4.27 (*qa*, $J=7$, 1H, S-CH-S); 4.25 (*qa*, $J=7$, 1H, S-CH-S); 2.53 (*s*, 3 H, $\text{S}_2\text{-CH}_3$); 2.14 (*s*, 3 H, S-CH $_3$); 1.70 (*d*, $J=7$, 3 H, CH $_3$); 1.60 (*d*, $J=7$, 3 H, CH $_3$). - MS.: 214 (M^+ , 0), 75 (100), 135 (23), 59 (22), 45 (20), 47 (12), 60 (11), 79 (7), 107 (5).

Threo-4,6-dimethyl-2,3,5,7-tetrathiooctane ((\pm))-**1b**. $^1\text{H-NMR}$.: 4.12 (*qa*, $J=7$, 1H, S-CH-S); 4.06 (*qa*, $J=7$, 1H, S-CH-S); 2.52 (*s*, 3 H, $\text{S}_2\text{-CH}_3$); 2.21 (*s*, 3 H, S-CH $_3$); 1.70 (*d*, $J=7$, 3 H, CH $_3$); 1.66 (*d*, $J=7$, 3 H, CH $_3$). - MS.: 214 (M^+ , 0), 75 (100), 59 (30), 45 (29), 135 (20), 47 (16), 60 (16), 79 (10), 107 (6).

The authors are indebted to Dr. M. Hřivnáč and Mrs. L. Šýkora for various preparative GC. and to Dr. P. Schudel for his stimulating interest in this work.

REFERENCES

- [1] M. Hřivnáč & L. Šýkora-Čechová, private communication.
- [2] H. W. Brinkman, H. Copier, J. J. M. de Leuw & S. B. Tjan, *J. agr. Food Chemistry* 20, 177 (1972).
- [3] P. Dubs & R. Stüssi, *Helv. 61*, 2351 (1978).
- [4] S. S. Chang, C. Hirai, B. R. Reddy, K. O. Herz, A. Kato & G. Sipma, *Chemistry & Ind.* 1968, 1639.
- [5] R. A. Wilson, C. J. Mussinan, I. Katz & A. Sanderson, *J. agric. Food Chemistry* 21, 873 (1973).
- [6] S. B. Tjan, J. C. Haakman, C. J. Teunis & H. G. Peer, *Tetrahedron* 28, 3489 (1972).